

Available online at www.sciencedirect.com



Tetrahedron Letters 47 (2006) 2431-2434

Tetrahedron Letters

Preparation of new amide-linked pseudodisaccharides by the carboxymethylglycoside lactone (CMGL) strategy

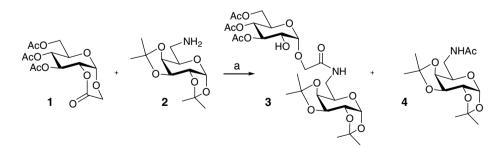
Alexandra Le Chevalier, Ronan Pierre, Rami Kanso, Stéphane Chambert, Alain Doutheau and Yves Queneau*

Laboratoire de Chimie Organique, UMR 5181 CNRS-UCBL-INSA, Institut National des Sciences Appliquées, 20 avenue A. Einstein, 69621 Villeurbanne, France

Received 26 October 2005; revised 25 January 2006; accepted 30 January 2006 Available online 20 February 2006

Abstract—6-Aminodeoxy sugars react with carboxymethyl 3,4,6-tri-O-acetyl- α -D-glucopyranoside 2-O-lactone (α -CMGL) to provide the corresponding new amide-linked glucose/glucose or glucose/galactose pseudodisaccharides in good yields. The strategy is extended to the synthesis of an amide-linked sugar/nucleoside adduct. © 2006 Elsevier Ltd. All rights reserved.

In recent years, considerable interest has been devoted to the design and synthesis of amide-linked polysaccharidic structures.¹⁻⁴ The assembly of modified carbohydrates bearing both an amino and a carboxylate function, called sugar amino acids (SAAs), using carbohydrate and peptide chemistry has provided compounds of biological interest. For example, antiviral activity against HIV and inhibition of sialyl Lewis x-dependant cell adhesion have been found for some sulfated amide-linked oligomers.^{5,6} SAAs oligomers have also been used for their ability to mimic peptides and, in some cases, have shown to adopt well-defined secondary conformations.⁷⁻⁹ We have recently reported that the readily available carboxymethyl 3,4,6-tri-O-acetyl- α -D-glucopyranoside 2-O-lactone 1 (α -CMGL) was a useful synthon for anchoring a carbohydrate moiety on several alcohols and amines via the selective opening of the lactone ring.^{10,11} We report here, the synthesis of new amidelinked pseudodisaccharides such as 3 using the opening of α -CMGL by various aminodeoxy sugars (Scheme 1). The length of the four-atom inter-glycosidic linkage being close to that of a monosaccharide unit, such derivatives can also be considered as trisaccharides mimetics. In the literature, such oligosaccharide analogues having non-glycosidic spacers have been reported. For example,



Scheme 1. Reagents and conditions: THF, rt, 18 h, 90% (3).

^{*} Corresponding author. Tel.: +33 (0)4 72 43 61 69; fax: +33 (0)4 72 43 88 96; e-mail: yves.queneau@insa-lyon.fr

^{0040-4039/\$ -} see front matter © 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2006.01.146

linear or cyclic oligosaccharides with a four-carbon rigid connection were constructed,^{12–15} and a compound bearing a six-atom acyclic spacer between two glucose residues has been synthesized and was found to be a competitive inhibitor of the hydrolysis of *p*-nitrophenyl α -maltotriose by porcine alpha-amylase.¹⁶

The ring opening of lactone 1 was first studied using the protected 6-aminodeoxy galactopyranose 2^{17} as a nucleophile for obtaining compound 3 (Scheme 1). NMR spectroscopy showed the typical pattern of carboxymethylglycoside adducts, including H(2) and H(3), respectively, at 3.83 and 5.33 ppm.¹⁸ Different reaction conditions were employed varying solvent, concentration and relative stoichiometry of the reactants. Unlike what is observed for the opening of α -CMGL 1 with alcohols,¹¹ it was noted that the reaction with amine 2 did not require the presence of DMAP. On the contrary, it was shown that its use led to an increase in the vield of the undesired N-acetylated derivative 4^{19} arising from competitive intermolecular O-to-N acetyl exchange. The obtained yields showed that the best result is achieved when the reactants are placed in THF with a slight excess of lactone 1. In these conditions, only trace amounts of 4 were detected by TLC.

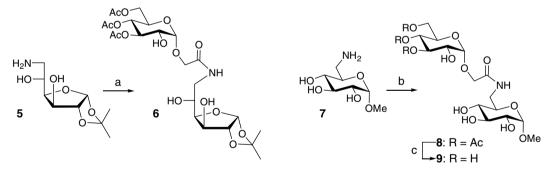
The reaction of lactone 1 with two partially protected 6-aminodeoxy sugars 5^{20} and 7^{21} was then studied (Scheme 2). For solubility reasons, reaction with compound 7 was performed in anhydrous DMF. Reactions yielded the partially protected pseudodisaccharides 6 and 8, with the expected selectivity for the amine versus the alcohol functions. Deprotection of the obtained

amide-linked pseudodisaccharide **8** was carried out using standard Zemplén conditions leading to the free pseudodisaccharide **9** (Scheme 2).

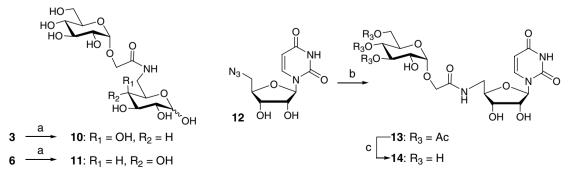
In the case of compounds **3** and **6**, both isopropylidene and acetyl groups were removed simultaneously using a 0.5 M HCl solution at 50 °C to give access to the corresponding fully unprotected compounds **10** and **11** (Scheme 3),²³ showing the reasonably good stability of the α -carboxymethyl linkage of such derivatives under acidic conditions.

The same strategy was finally applied to the synthesis of an amide-linked sugar/nucleoside adduct. Thus, 5'-deoxy-5'-azidouridine 12^{24} was reduced (H₂, Pd/C) to the corresponding 5'-deoxyamino analogue, which was reacted without further purification with lactone 1 in DMF to furnish, after removal of the acetyl groups, the amide 14 (Scheme 3). No protection of the uracil moiety was needed.

In summary, we have demonstrated that the opening of carboxymethyl 3,4,6-tri-O-acetyl- α -D-glucopyranoside 2-O-lactone (α -CMGL) with deoxyamino sugars constitutes an efficient strategy for constructing amide-linked pseudodisaccharides. This reaction has been applied to the synthesis of new glucose–glucose, glucose–galactose and glucose–uridine adducts. We are now studying the scope of this reaction with respect to other nucleophilic species in order to obtain more elaborated glycoconjugate analogues similar to compound 14, which could potentially mimic the glucosyltransferases substrate (UDP-Glc). Extension to carboxymethyl glycoside



Scheme 2. Reagents and conditions: (a) THF, rt, 18 h, 90%; (b) DMF, rt, 15 h, 87%; (c) MeONa/MeOH, rt, 1 h, 83%.²²



Scheme 3. Reagents and conditions: (a) 0.5 M HCl, dioxane/H₂O, 50 °C, 12 h, 80%; (b) (i) H₂, Pd/C, EtOH, rt, 15 h, (ii) 1, DMF, rt, 70% overall; (c) MeONa/MeOH, rt, 1 h, 82%.²⁵

lactones derived from other mono or oligosaccharides is also under investigation.

Acknowledgements

Financial support from CNRS and MENESR is gratefully acknowledged as well as a grant to R.P. from CNRS and TEREOS.

References and notes

- 1. Gruner, S. A. W.; Locardi, E.; Lohof, E.; Kessler, H. Chem. Rev. 2002, 102, 491-514.
- 2. Schweizer, F. Angew. Chem., Int. Ed. 2002, 41, 230-253.
- 3. Wessel, H. P. In Carbohydrates in Chemistry and Biology; Ernst, B., Hart, G. W., Sinaÿ, P., Eds.; Wiley-CCH: Weinheim, Germany, 2000; Vol. 1, pp 565-585.
- 4. Wessel, H. P. In GlycoScience: Chemistry and Chemical Biology, Part II; Fraser-Ried, B. O., Tatsuta, K., Thiem, J., Eds.; Springer: Heidelberg, Germany, 2001; pp 2725-2752
- 5. Suhara, Y.; Ichikawa, M.; Hildreth, J. E. K.; Ichikawa, Y. Tetrahedron Lett. 1996, 37, 2549-2552.
- Suhara, Y.; Yamaguchi, Y.; Collins, B.; Schnaar, R. L.; 6 Yanagishita, M.; Hildreth, J. E. K.; Shimada, I.; Ichikawa, Y. Bioorg. Med. Chem. 2002, 10, 1999-2013.
- 7. Graf von Roedern, E.; Lohof, E.; Hessler, G.; Hoffmann, M.; Kessler, H. J. Am. Chem. Soc. 1996, 118, 10156-10167.
- Gregar, T. Q.; Gervay-Hague, J. J. Org. Chem. 2004, 69, 8. 1001-1009.
- 9. Claridge, T. D. W.; Long, D. D.; Baker, C. M.; Odell, B.; Grant, G. H.; Edwards, A. A.; Tranter, G. E.; Fleet, G. W.; Smith, M. D. J. Org. Chem. 2005, 70, 2082-2090.
- 10. Trombotto, S.; Bouchu, A.; Descotes, G.; Queneau, Y. Tetrahedron Lett. 2000, 41, 8273-8277.
- 11. Trombotto, S.; Danel, M.; Fitremann, J.; Bouchu, A.; Queneau, Y. J. Org. Chem. 2003, 68, 6672-6678.
- 12. Alzeer, J.; Cai, C.; Vasella, A. Helv. Chim. Acta 1995, 78, 242-264.
- 13. Alzeer, J.; Vasella, A. Helv. Chim. Acta 1995, 78, 1219-1237.
- Brüli, R.; Vasella, A. Helv. Chim. Acta 1997, 80, 1027-14. 1052.
- 15. Brüli, R.; Vasella, A. Helv. Chim. Acta 1997, 80, 2215-2237.
- 16. Jegge, S.; Lehmann, J. Carbohydr. Res. 1984, 133, 247-254.
- 17. Reitz, A. B.; Tuman, R. W.; Marchione, C. S.; Jordan, A. D., Jr.; Bowden, C. R.; Maryanoff, B. E. J. Med. Chem. **1989**, 32, 2110–2116.
- 18. Compound **3**: ¹H NMR (500 MHz, CDCl₃): δ 7.81 (d, J = 7.3 Hz, 1H), 5.55 (d, J = 5 Hz, 1H), 5.33 (t, J = 9.9 Hz, 1H), 5.01 (t, J = 9.4 Hz, 1H), 4.81 (d, J =3.6 Hz, 1H), 4.58 (dd, J = 7.9 Hz, J = 2.4 Hz, 1H), 4.34 (dd, J = 5.0 Hz, J = 2.4 Hz, 1H), 4.22–4.34 (m, 3H), 4.05–4.12 (m, 2H), 3.94 (ddd, J = 13.9 Hz, J = 9.2 Hz, J = 2.0 Hz, 1H), 3.83 (dd, J = 9.9 Hz, J = 3.6 Hz, 1H), 3.79 (m, 1H), 3.16 (ddd, J = 12.6 Hz, J = 7.1 Hz, J = 2.0 Hz, 1H), 2.07 (s, 3H), 2.07 (s, 3H), 2.02 (s, 3H), 1.46 (s, 3H), 1.42 (s, 6H), 1.32 (s, 3H); ¹³C NMR (126 MHz, CDCl₃): δ 171.2, 171.1, 170.2, 169.5, 110.2, 109.5, 99.8, 96.7, 72.7, 71.9, 70.9, 70.9, 70.2, 68.5, 68.4, 68.4, 67.9, 62.3, 39.6, 26.4, 26.3, 25.2, 24.7, 21.3, 21.1, 21.0 +78 (c 1.0, CH₂Cl₂). HRMS (ES) calcd for $[M+Na]^+$ $|\alpha|_{\rm D}^{20}$ 628.2217, found 628.2225. Anal. Calcd for C₂₆H₃₉NO₁₅:

C, 51.57; H, 6.49; N, 2.31. Found: C, 51.27; H, 6.68; N, 2.22.

- 19. Compound 4 exhibited same spectroscopic data as those reported: Streicher, B.; Wünsch, B. Carbohydr. Res. 2003, 338. 2375-2385.
- 20. Cramer, F.; Otterbach, H.; Springmann, H. Chem. Ber. 1959, 92, 384-391.
- 21. Reddington, M. V. J. Chem. Soc., Perkin Trans. 1 1998, 143-147.
- 22. Compound 6: ¹H NMR (500 MHz, CDCl₃): δ 8.26 (t, J = 8.3 Hz, 1H), 5.92 (d, J = 3.0 Hz, 1H), 5.30 (t, J = 9.6 Hz, 1H), 5.01 (t, J = 9.9 Hz, 1H), 4.93 (d, J = 3.3 Hz, 1H), 4.54 (d, J = 3.4 Hz, 1H), 4.34 (m, 3H), 4.06-3.93 (m, 5H), 3.80-3.60 (m, 1H), 3.74 (dd, J = 9.6, J = 3.3 Hz, 1H), 3.33-3.28 (m, 1H), 2.09 (s, 3H), 2.08 (s, 3H), 2.03 (s, 3H), 1.47 (s, 3H), 1.30 (s, 3H); ¹³C NMR (126 MHz, CDCl₃): 171.7, 171.5, 171.1, 170.1, NMR (120 μ 12, CEC3), 1711, 1722, 1723, 10.5, 69.5, 69.5, 111.9, 105.4, 99.9, 85.5, 81.1, 75.2, 73.2, 70.5, 69.5, 69.5, 69.5, 68.4, 67.8, 62.3, 43.8, 27.0, 26.4, 21.3, 21.2, 21.1. $[\alpha]_{\rm D}^{20}$ +95 $(c 1.0, CH_2Cl_2)$. HRMS (ES) calcd for $[M+Na]^+$ 588.1904, found 588.1909. Anal. Calcd for C₂₃H₃₅NO₁₅: C, 48.85; H, 6.24; N, 2.48. Found: C, 48.55; H, 6.35; N, 2.41. Compound 8: ¹H NMR (300 MHz, MeOD): δ 5.23 (t, J = 9.8 Hz, 1H), 4.95 (t, J = 9.8 Hz, 1H), 4.94 (d, J = 3.9 Hz, 1H), 4.68 (d, J = 3.7 Hz, 1H), 4.26 (d, J = 15.5 Hz, 1H), 4.25 (dd, J = 12.2 Hz, J = 4.8 Hz, 1H), 4.09 (d, J = 15.5 Hz, 1H), 4.07 (dd, J = 12.2 Hz, J = 2.3 Hz, 1H), 4.02 (ddd, J = 10.3 Hz, J = 4.8 Hz, J = 2.3 Hz, 1H), 3.76 (dd, J = 9.8 Hz, J = 3.9 Hz, 1H), 3.66-3.56 (m, 4H), 3.42 (dd, J = 9.6 Hz, J = 3.7 Hz, 1H), 3,16 (t, J = 8.9 Hz, 1H), 3.40 (s, 3H), 2.05 (s, 6H), 2.01 (s, 3H): ¹³C NMR (75 MHz, MeOD): 172.4, 172.2, 172.0, 171.5, 101.3, 100.6, 74.9, 74.3, 73.5, 73.3, 71.3, 71.0, 70.0, 69.4, 68.0, 63.3, 55.8, 41.1, 20.9, 20.7, 20.7. $[\alpha]_{\rm D}^{20}$ +119 (*c* 1.0, MeOH). HRMS (ES) calcd for $[M+Na]^+$ 562.1748, found 562.1749. Anal. Calcd for C₂₁H₃₃NO₁₅: C, 46.75; H, 6.17; N, 2.60. Found: C, 46.30; H, 6.52; N, 2.51. Compound 9: ¹H NMR (500 MHz, MeOD): δ 4.83 (d, J = 3.7 Hz, 1H), 4.67 (d, J = 3.7 Hz, 1H), 4.23 (d, J = 15.7 Hz, 1H), 4.04 (d, J = 15.7 Hz, 1H), 3.81 (dd, J = 11.9 Hz, J = 2.3 Hz, 1H), 3.68 (dd, J = 11.9 Hz,J = 5.4 Hz, 1H), 3.67 (t, J = 9.5 Hz, 1H), 3.63–3.52 (m, 5H), 3.48 (dd, J = 9.7 Hz, J = 3.7 Hz, 1H), 3.42 (dd, J = 9.6 Hz, J = 3.7 Hz, 1H), 3.40 (s, 3H), 3.32 (t, J = 9.3 Hz, 1H), 3.17 (t, J = 9.2 Hz, 1H); ¹³C NMR J = 9.3 Hz, 1H), 3.17 (t, J = 9.2 Hz, 1H), (75 MHz, MeOD): 172.6, 101.2, 101.0, 74.9, 74.8, 74.4, 75 MHz, 72.2, 71.5, 71.4, 67.7, 62.5, 55.8, 41.1. $[\alpha]_D^{20}$ +155 (c 1.0, MeOH). HRMS (ES) calcd for $[M+Na]^+$ 436.1431, found 436.1432.
- 23. Compounds 10 and 11 were identified by mass spectroscopy and characteristic ¹H NMR patterns have been assigned. Compound 10: ¹H NMR (300 MHz, D_2O): δ 5.24 (d, J = 3.6 Hz, H-1' α), 4.98 (d, J = 3.6 Hz, H-1), 4.56 (d, J = 7.8 Hz, H-1' β), 4.28 and 4.13 (d, J = 15.6 Hz, AB system, H-7), $\alpha:\beta = 4:6. \ [\alpha]_D^{20} + 102$ (c 1.0, H₂O); HRMS-LSIMS calcd for [M+H]⁺ 400.1455, found 400.1457. Compound 11: ¹H NMR (500 MHz, D_2O): δ 5.25 (d, J = 3.9 Hz, H-1' α), 5.02 (d, J = 3.8 Hz, H-1), 4.68 (d, J = 8.1 Hz, H-1' β), 4.33 and 4.18 (d, J = 15.6 Hz, AB system, H-7), $\alpha:\beta = 3:7. [\alpha]_D^{20} +90 (c \ 0.7, H_2O);$ HRMS-LSIMS calcd for [M+H]⁺ 400.1455, found 400.1459. 24. Yeager, A. R.; Finney, N. S. J. Org. Chem. **2004**, 69, 613–
- 618.
- 25. Compound 13: ¹H NMR (300 MHz, CD₃OD): δ 7.66 (d, J = 8.1 Hz, 1H), 5.76 (d, J = 8.1 Hz, 1H), 5.75 (d, J = 4.4 Hz, 1H), 5.33 (t, J = 9.6 Hz, 1H), 4.96 (t, J = 9.7 Hz, 1H), 4.95 (d, J = 3.7 Hz, 1H), 4.31–4.22 (m, 3H), 4.13–4.02 (m, 5H), 3.77 (dd, J = 9.7 Hz, J =

3.7 Hz, 1H), 3.68 (dd, J = 14.1 Hz, J = 6.0 Hz, 1H), 3.57 (dd, J = 14.1 Hz, J = 3.6 Hz, 1H), 2.05 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H); ¹³C NMR (50 MHz, CD₃OD): 172.4, 172.4, 172.1, 171.4, 166.0, 152.3, 143.2, 103.1, 100.6, 92.3, 83.8, 74.5, 74.3, 72.3, 70.9, 69.9, 69.2, 68.3, 63.3, 41.7, 20.9, 20.7, 20.7. [α]_D²⁰ +80 (c 1.0, MeOH). LSIMS [M+H]⁺ 590, [M+Na]⁺ 612. Anal. Calcd for C₂₃H₃₁NO₁₅·0.33H₂O: C, 46.39; H, 5.36; N, 7.06. Found: C, 46.39; H, 5.55; N, 7.44. Compound **14**: ¹H NMR (300 MHz, D₂O): δ 7.69 (d,

 $J = 8.1 \text{ Hz}, 1\text{H}, 5.89 \text{ (d, } J = 8.1 \text{ Hz}, 1\text{H}, 5.79 \text{ (d, } J = 4.2 \text{ Hz}, 1\text{H}), 4.96 \text{ (t, } J = 3.6 \text{ Hz}, 1\text{H}), 4.39 \text{ (t, } J = 4.2 \text{ Hz}, 1\text{H}), 4.28 \text{ (d, } J = 15.6 \text{ Hz}, 1\text{H}), 4.16-4.08 \text{ (m, 3H)}, 3.86-3.63 \text{ (m, 6H)}, 3.60 \text{ (dd, } J = 9.9, J = 3.6 \text{ Hz}, 1\text{H}); 3.42 \text{ (t, } J = 9.0 \text{ Hz}, 1\text{H}); ^{13}\text{C} \text{ NMR} \text{ (50 MHz}, D_2\text{O}): 172.8, 166.7, 152.0, 142.9, 102.8, 99.4, 91.3, 82.2, 73.5, 73.3, 72.8, 71.6, 71.0, 69.9, 66.9, 60.9, 40.7. } [\alpha]_D^{2} + 82 \text{ (c } 1.0, \text{ MeOH)}. \text{HRMS-LSIMS calcd for } [\text{M+H}]^+ 464.1517, \text{ found } 464.1513.$